## **Course Outcomes Guide (COG)**

Course Title: CHM 103 Date: June 2019

Course Team: Veronica Stein, Chris Nelling

## **Expected Learning Outcomes**

- 1. Apply quantitative thinking processes and reasoning skills to physical laws, stoichiometry, and atomic and molecular structure.
- 2. Communicate core course concepts in writing while using appropriate technology
- 3. Solve quantitative chemistry problems and demonstrate reasoning clearly and completely. Integrate multiple ideas in the problem solving process. Check results to make sure they are physically reasonable.
- 4. Collect, analyze, and evaluate empirical data to substantiate chemical concepts.
- 5. Access, process, analyze and synthesize scientific information.
- 6. Relate chemical concepts to real life scenarios

**Assessment** (How do or will students demonstrate achievement of each outcome? Please attach a copy of your assessment electronically.)

- MasteringChemistry, which is an online homework program, is used to assess applying quantitative thinking process and reasoning skills, solving quantitative chemistry problems and demonstrate reasoning clearly and completely.
- Exams, consisting of a combination of multiple-choice, short answer and problem solving questions, are given in the lecture and lab sections of the course. The exams access critical thinking skills and analyze and synthesis of scientific information.
- A Nationalized Final Exam written by the American Chemical Society (ACS) for the first semester of General Chemistry is used as the final exam for CHM 103.

**Validation** (What methods have you used or will you use to validate your assessment?) We compare our students to the national average of the ACS exam. This exam covers material from chapters 1 through 10 in the *Chemistry, A Molecular Approach, 4<sup>th</sup> edition*, by Tro.

**Results** (What do your assessment data show? If you have not yet assessed student achievement of your learning outcomes, when is assessment planned?)

Our students typically achieve the mean or higher for the national ACS exam 63% of the time using the 2009 version of the exam and 45% with the 2015 version of the exam.

CHM 103										
Semester	n	mean								
05/FA	51	33.7								
06/SP	13	36,5								
06/SU	5	56.4								
06/FA	42	42.5								
07/SP	21	37.6								

CHM 103									
07/SU	7	42.57							
07/FA	53	38.21							
08/SP	19	40.05							
08/SU	7	43.29							
08/FA	54	42.67							
09/SP	25	39.08							
09/Su	11	38.73							
National 2002	2616 from 32	41.03							
version	colleges								

out of 70 questions

CHM 103									
Semester	n	mean							
09/FA	58	41.1							
10/SP	23	34.6							
10/SU	10	46.7							
10/FA	74	39.8							
11/SP	31	35.7							
11/SU	23	41.0							
11/FA	52	40.9							
12/SP	34	37.1							
12/SU	18	37.6							
12/FA	51	41.7							
13/SP	27	33.1							
13/SU	19	35.79							
13/FA	57	42.28							
14/SP	22	33.82							
14/SU	14	34.71							
14/FA	65	41.15							
15/SP	17	43.41							
15/SP	18	30.94							
15/SU	12	37.83							
15/FA	56	38.66							
16/SP	17	39.65							
16/SP	14	37.29							
16/SU	19	33.8							
16/FA	49	38.4							
National	3827	37.13							
2009	from 34								

out of 70 questions

CHI	<i>I</i> 103	
Semester	n	mean
17/SP	11	33.3
17/SP	21	27.8
17/SU	14	40.1
17/FA	56	37.0

version

colleges

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	CH	IM 103	
	18/SP	15	42.2
	18/SP	19	32.4
	18/SU	13	41.3
	18/FA	71	38.4
	19/SP	11	44.6
	19/SP		
	Motional	72.47	20.7

National	7347	39.7	out of 70
2015	From 47		
version	institution	S	

out of 70 questions

Follow-up (How have you used or how will you use the data to improve student learning?)

I made an Excel spreadsheet table for our adjuncts and me, which correlates the item analysis of the exam to chemistry concepts. Using this we can see which concepts we need to develop better curriculum (practice problems, labs, additional material) for students. To help students associate chemistry concepts used in lecture with laboratory experiments we will develop videos demonstrating laboratory techniques used in various experiments. Also, we will show how these laboratory techniques are used in industry or in the kitchen.

**Budget Justification** (What resources are necessary to improve student learning?)

A newer version of the ACS exams should be purchased to keep current with the content in the newer versions of the text books.

Course: CHM 103 SLOA Data Faculty Team: V Stein

Course. Criv	SLOA Data						racuity realli. V Stelli					
	SU 2009	FA 2009	SP 2010	SU 2010	FA 2010	SP 2011	SU 2011	FA 2011	SP 2012	SU 2012	FA 2012	SP 2013
# Active students	13	74	30	17	93	37	24	75	39	19	80	45
%W	7.7	9.5	13.3	23.5	10.8	8.1	0	17.3	10.3	0	26.3	4.4
*% walk-away Fs No final exam/grade = F	0	8.1	7.7	15.4	7.2	8.8	4.2	8.1	2.9		13.3	
% Success (A,B,C)	66.7	59.2	75.9	37.5	63.3	59.5	87.5	55.7	66.7	73.7	50.6	75.0
Mean Common Lab Practical Score												
Common Comprehensive Final Exam Score (70 questions)	38.7	41.1	34.6	46.7	39.8	35.7	41.0	40.9	37.1	37.6	41.7	34.06
Mean course grade	2.09	1.91	2.64	2.17	2.04	2.12	2.17	2.07	2.46	2.37	1.97	2.33
% Gen Ed Assessment Score											74.3	73.0
Item Analysis Weakest Content Areas												

<sup>\*%</sup> Walk-away Fs = did not take the final exam and received a grade of F.

Course: CHM 103 SLOA Data Faculty Team: V Stein

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	SU 2013	FA 2013	SP 2014	SU 2014	FA 2014	SP 2015	SU 2015	FA 2015	SP 2016	SU 2016	FA 2016	SP 2017
# Active students	22	72	29	17	80	58	15	59	41	22	71	36
%W	9.1	19.5	6.9	11.8	15.0	12.1	6.7	5.1	17.1	13.6	15.5	2.8
*% walk-away Fs No final exam/grade = F		1.4			1.1			5.1			9.8	
% Success (A,B,C)	68.2	55.6	50.0	56.3	61.0	58.9	73.3	61.0	61.0	68.2	56.5	58.3
Mean Common Lab Practical Score												
Common Comprehensive Final Exam Score (70 questions)	35.8	42.3	33.8	34.71	41.2	37.8	37.8	38.7	38.6	33.8	38.5	29.7
Mean course grade	2.75	2.17	1.85	2.57	2.12	2.06	2.62	2.11	2.21	2.26	1.90	1.97
% Gen Ed Assessment Score	77.4	86.1	75.9	78.3	63.8	75.3	83.3	83.0	80.6	75.3	77.3	76.2
Item Analysis Weakest Content Areas												

<sup>\*%</sup> Walk-away Fs = did not take the final exam and received a grade of F.

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	DEGIT Data						racuity reall. V Stelli						
	SU	FA	SP	SU	FA	SP	SU	FA	SP	SU	FA	SP	
	2017	2017	2018	2018	2018	2019	2019	2019	2020	2021	2021	2022	
# Active students	15	68	44										
%W	6.7	8.8	6.8										
*% walk-away													
Fs													
No final exam/grade = F													
% Success (A,B,C)	73.3	59.1	66.7										
Mean Common Lab Practical Score													
Common Comprehensive Final Exam Score (70 questions)	40.1	37.1	36.7	41.3	38.4	44.6**							
Mean course grade	2.36	1.93	2.15										
% Gen Ed Assessment Score	94.3	85.0	77.0	81.5	83.5	88.2**							
Item Analysis Weakest Content Areas				**Need Chris Nelling's data ***see below									

<sup>\*%</sup> Walk-away Fs = did not take the final exam and received a grade of F.

## **Weakest Content Areas**

\*\*\*Using item analysis on the ACS exam, the following concepts need improvements: electron transitions and energy associated with this transition, orbital diagrams, formal charges resonance with Lewis structures polarity of a molecule, Limiting and excess reactants in a reaction, Kinetic molecular theory in gas laws, using bond dissociation energies to determine heat of reaction, intermolecular forces (chapter 11 material, which is not covered until CHM 104).